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Activity 1

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different salts
to water, then
watch them
dissolve and
achieve a
dynamic
equilibrium with
solid
precipitate.
Compare the

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number of ions
in solution for
highly soluble
NaCl to other
slightly soluble
salts. Relate
the charges on
ions to the
number of ions
in the formula
of a salt.
Calculate K_{sp}
values.

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Definition of
Solubility
Solubility is
the ability of a
solid, liquid,
or gaseous

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simulation is
called
concentration
and is available
online on the
PhET website.

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What happens
when sugar and
salt are added
to water? Pour
in sugar, shake
in salt, and
evaporate water
to see the
effects on
concentration

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conductivity.
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Zoom in to see
how different
sugar and salt
compounds
dissolve. Zoom
in again to
explore the role
of water.

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Visualizing
ionic formulas
using Salts and
Solubility

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Activity 1

Learning Goals

Students will be
able to: •

Determine the
chemical formula
by observation

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of ionic ratios
in solutions •
Relate the
simulation scale
to real lab
equipment
through
illustration and
calculations

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Definition of
Solubility.

Solubility is
the ability of a
solid, liquid,
or gaseous
chemical
substance
(referred to as
the solute) to
dissolve in
solvent (usually

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a liquid) and
form a
solution. The
solubility of a
substance

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a liquid) and
form a

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